This article was downloaded by: On: 22 January 2011 Access details: Access Details: Free Access Publisher Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



# Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713454007

# Three new flavonol triglycosides from Derris trifoliata

Lu-Rong Xu<sup>ab</sup>; Pei Zhou<sup>a</sup>; Yue-E Zhi<sup>a</sup>; Jun Wu<sup>c</sup>; Si Zhang<sup>c</sup> <sup>a</sup> School of Agriculture and Biology, Shanghai Jiao Tong University, Shanghai, China <sup>b</sup> School of Life Sciences, Shanghai University, Shanghai, China <sup>c</sup> South China Sea Institute of Oceanology, Chinese Academy of Sciences, Guangzhou, China

**To cite this Article** Xu, Lu-Rong , Zhou, Pei , Zhi, Yue-E , Wu, Jun and Zhang, Si(2009) 'Three new flavonol triglycosides from *Derris trifoliata*', Journal of Asian Natural Products Research, 11: 1, 79 – 84 **To link to this Article: DOI:** 10.1080/10286020802514598 **URL:** http://dx.doi.org/10.1080/10286020802514598

# PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.



## Three new flavonol triglycosides from Derris trifoliata

Lu-Rong Xu<sup>ac</sup>, Pei Zhou<sup>a</sup>\*, Yue-E Zhi<sup>a</sup>, Jun Wu<sup>b</sup> and Si Zhang<sup>b</sup>

<sup>a</sup>School of Agriculture and Biology, Shanghai Jiao Tong University, Shanghai, China; <sup>b</sup>South China Sea Institute of Oceanology, Chinese Academy of Sciences, Guangzhou, China; <sup>c</sup>School of Life Sciences, Shanghai University, Shanghai, China

(Received 24 April 2008; final version received 10 September 2008)

Three new flavonol triglycosides, kaempferol-3-O- $\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 6)$ - $\beta$ -D-glucopyranosyl- $(1 \rightarrow 3)$ - $\beta$ -D-glucopyranoside (1), quercetin-3-O- $\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 6)$ - $\beta$ -D-glucopyranosyl- $(1 \rightarrow 2)$ - $\beta$ -D-glucopyranoside (2), quercetin-3-O- $\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 6)$ - $\beta$ -D-glucopyranosyl- $(1 \rightarrow 2)$ - $\beta$ -D-glucopyranoside (3), together with the two known flavonol glycosides, kaempferol-3-O- $\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 6)$ - $\beta$ -D-glucopyranosyl- $(1 \rightarrow 2)$ - $\beta$ -D-glucopyranosyl- $(1 \rightarrow 6)$ - $\beta$ -D-glucopyranosyl- $(1 \rightarrow 2)$ - $\beta$ -D-glucopy

Keywords: Derris trifoliata; Leguminosae; flavonol triglycosides

#### 1. Introduction

Derris trifoliata (Leguminosae) is a woody climber distributed in the Mangrove in Southeast Asia, used popularly as poison for fish hunt and medical stimulant, antisasmodic and counter-irritant agents by local people [1]. Many flavonoids have been isolated from different species of Derris, including flavanones [2,3], flavones [4-6], isoflavone glycosides [7,8], rotenoids [9], chalcones [10], and aurones [11]. The present work on this plant has resulted in the isolation and structure elucidation of three new flavonol triglycosides (1-3), Figure 1), along with the two known flavonol glycosides, kaempferol-3-O-α-L-rhamnopyranosyl- $(1 \rightarrow 6)$ - $\beta$ -D-glucopyranoside (4) [12] and kaempferol-3-O- $\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 6)$ - $\beta$ -D-glucopyranosyl- $(1 \rightarrow 2)$ - $\beta$ -D-glucopyranoside (5) [13].

## 2. Results and discussion

Compound 1, yellow powder, was assigned the molecular formula C<sub>33</sub>H<sub>40</sub>O<sub>20</sub> from its  $[M - H]^{-}$  peak at m/z 755.2036 in the negative HR-SI-MS. Its UV spectrum exhibited characteristic absorption maxima of flavonols at  $\lambda_{max}$ 265 and 347 nm. The <sup>1</sup>H and <sup>13</sup>C NMR spectra showed the characteristic signals of flavonoid glycosides. The aglycone was identified as 3,5,7,4'-tetrahydroxyflavone (kaempferol) from the <sup>1</sup>H NMR spectrum with the two meta-coupled doublets at  $\delta$  6.19 (1H, d, J = 2.0 Hz, H-6) and 6.38 (1H, d, J = 2.0 Hz,H-8) for the A ring, and two ortho-coupled doublets at  $\delta$  8.02 (2H, d, J = 9.0 Hz, H-2', 6') and 6.89 (2H, d, J = 9.0 Hz, H-3', 5') for the B ring [14]. Acid hydrolysis indicated the existence of glucose and rhamnose moieties. One rhamnosyl moiety was indicated by the anomeric proton at  $\delta$  4.47 (1H, d, J = 1.0 Hz,

<sup>\*</sup>Corresponding author. Email: zhoupei@sjtu.edu.cn





Figure 1. Important HMBC correlations of compounds 1-3.

H-1<sup>////</sup>), the anomeric carbon at  $\delta$  102.3 (C-1<sup>////</sup>), and a methyl group at  $\delta$  17.9. The presence of two glucosyl moieties was apparent from two anomeric protons at  $\delta$  5.35 (1H, d, J = 7.5 Hz, H-1") and 4.75 (1H, d, J = 7.0 Hz, H-1") and the corresponding carbon signals at  $\delta$  101.1 (C-1''') and 104.6 (C-1'''). The large coupling constant of the two glucosyl anomeric proton signals  $(J = 7.5, 7.0 \,\mathrm{Hz})$  indicated that the two glucosyl moieties were both of β-configuration [15]. The rhamnosyl moiety was linked to the terminal of the outer glucosyl moiety from the significantly downshifted signal of C-6<sup>///</sup> ( $\delta$ 68.3) and HMBC correlations between H-1<sup>///</sup> and C-6<sup>*III*</sup>. The glucosyl moiety substituted by the rhamnosyl moiety was attached to the hydroxyl group at C-3" of the inner glucosyl moiety, according to the downshifted signal of C-3" at  $\delta$  82.3 and HMBC correlations between H-1<sup>*III*</sup> and C-3<sup>*II*</sup>. Linkage of the inner glucosyl moiety to C-3 of the aglycone was determined from the C-3 signal at  $\delta$  134.8, which was significantly upshifted comparing with the aglycone (kaempferol) [16]. Furthermore, the correlation between H-1" and C-3 was observed in HMBC spectrum. Thus, the structure of compound 1 was established as kaempferol-3-O- $\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 6)$ - $\beta$ -D-glucopyranosyl- $(1 \rightarrow 3)$ - $\beta$ -D-glucopyranoside.

Compound 2 was assigned the molecular formula C<sub>33</sub>H<sub>40</sub>O<sub>21</sub> from the negative HR-SI-MS spectrum showing a  $[M - H]^-$  peak at m/z 771.2001. The <sup>1</sup>H and <sup>13</sup>C NMR signals revealed a quercetin 3-O-triglycoside with the completely same sugar moieties as those of compound 1. The quercetin (3,5,7,3',4'-pentahydroxyflavone) aglycone was indicated by the two meta-coupled signals at  $\delta$  6.19 (1H, d, J = 2.0 Hz, H-6) and 6.37 (1H, d, J = 2.0 Hz, H-8) for the A ring and an ABX pattern system for the B ring:  $\delta$  7.65 (1H, d, J = 2.0 Hz, H-2'), 6.88 (1H, d, J = 8.5 Hz, H-5'), and 7.54 (dd, 1H, J = 2.0, 8.5 Hz, H-6' [16]. Thus, the structure of 2 was elucidated as quercetin-3-O- $\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 6)$ - $\beta$ -D-glucopyranosyl- $(1 \rightarrow 3)$ - $\beta$ -D-glucopyranoside.

Compound **3** was assigned the molecular formula  $C_{33}H_{40}O_{21}$  from the negative HR-SI-MS spectrum showing a  $[M - H]^-$  peak at m/z 771.2006. The <sup>1</sup>H and <sup>13</sup>C NMR signals indicated a quercetin 3-*O*-triglycoside similar to compound **2**, but the outer glucosyl moiety was attached to the hydroxyl group at C-2" of the inner glucosyl moiety rather than to OH (C-3") as in compound **2**, which was apparent from the downfield shift of C-2" at  $\delta$  80.2 together with the upfield shift of C-1" at  $\delta$  101.0 [14]. The attachment was further confirmed by HMBC correlation between H-1<sup>III</sup> and C-2". Consequently, the structure of compound **3** was deduced as quercetin-3-*O*- $\alpha$ -L-rhamnopyranosyl-(1  $\rightarrow$  6)- $\beta$ -D-glucopyranosyl-(1  $\rightarrow$  2)- $\beta$ -D-glucopyranoside.

#### 3. Experimental

#### 3.1 General experimental procedures

Optical rotations were measured on a Perkin-Elmer mode 241 polarimeter. The UV spectra were recorded on a Kontron Uvikon-860 spectrophotometer. The IR spectra were taken on a Bruker Equinox 55. The NMR spectra were recorded on a Bruker DRX-500 (500 MHz for <sup>1</sup>H and 125 MHz for <sup>13</sup>C, TMS as internal standard). HR-SI-MS spectra were obtained on a Bruker Daltonics, Inc.; FT-ICRMS APEX II. Spectrometer in the negative-ion mode. Preparative HPLC was performed on a system consisting of Waters 600 pump, a 600 controller, a 996 photodiode array detector, and an ODS column  $(250 \times 20 \text{ mm i.d.}, \text{YMC})$ . Column chromatography was performed over silica gel (200-300 mesh; Qingdao Mar. Chem. Ind. Co. Ltd, Qingdao, China), octadecylsilyl silica gel (80-100 µm; Shimadzu, Kyoto, Japan), Sephadex LH-20 gel (Pharmacia, Uppsala, Sweden), and D<sub>101</sub> macroporous resin (Tianjin Chem. Ind. Co. Ltd, Tianjin, China).

#### 3.2 Plant material

Aerial parts of *D. trifoliata* were collected from Tielugang, Sanya City, Hainan Province in October 2002, and authenticated by Dr Si Zhang. A voucher specimen (No. GKLMMM 003) is deposited at the Herbarium of South China Sea Institute of Oceanology. L.-R. Xu et al.

## 3.3 Extraction and isolation

The dry aerial parts (10 kg) of *D. trifoliata* were extracted thrice with 95% EtOH at 80°C, then thrice with 50% EtOH. After evaporation of the solvents under reduced pressure, the residues (1.5 kg) were suspended in  $H_2O$  and extracted with petroleum ether, ethyl acetate, and *n*-butanol successively.

The *n*-butanolic extracts from 95 and 50% EtOH were combined and subjected to a column of D<sub>101</sub> macroporous resin, eluted successively with H<sub>2</sub>O, 30% EtOH, and 60% EtOH. Eluate from 30% EtOH (30g) was chromatographed on a silica gel column using CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O gradient (9:1:0, 6:1:0, 6:2:0, 6:4:0, 6:4:0.5, and 6:4:1) to give five fractions. Fractions 1-4 were subjected to Rp-18 (MeOH-H<sub>2</sub>O in gradient) and Sephadex LH-20 (MeOH-H<sub>2</sub>O in gradient) repeatedly, yielding four subfractions that were followed by purification on preparative HPLC: subfraction 1 yielded compound 4  $(25 \text{ mg}; \text{ flow rate } 10 \text{ ml/min}, \text{ MeOH}-H_2\text{O})$ 32:78); subfraction 2 yielded compounds 5 (43 mg) and 1 (43 mg; flow rate 8 ml/min,MeCN-H<sub>2</sub>O 12:88); and subfraction 4 yielded compounds 3 (38 mg) and 2 (36 mg; flow rate 8 ml/min, MeCN-H<sub>2</sub>O 12:88).

## 3.3.1 Compound 1

Yellow powder;  $[\alpha]_D^{25} - 67 (c = 0.38, \text{MeOH})$ . UV  $\lambda_{\text{max}}$  (MeOH) nm: 265 and 347. IR  $\nu_{\text{max}}$  (KBr) cm<sup>-1</sup>: 3361, 2923, 1659, 1609, 1499, 1451, 1358, 1305, 1277, 1211, 1177, and 1071. <sup>13</sup>C NMR spectral data (CD<sub>3</sub>OD, Table 1). <sup>1</sup>H NMR spectral data (CD<sub>3</sub>OD, Table 2). HR-SI-MS (negative) m/z: 755.2036 [M – H]<sup>-</sup> (calcd for C<sub>33</sub>H<sub>39</sub>O<sub>20</sub>, 755.2037).

#### 3.3.2 Compound 2

Yellow powder;  $[\alpha]_D^{25} - 70 (c = 0.67, MeOH)$ . UV  $\lambda_{max}$  (MeOH) nm: 255 and 353. IR  $\nu_{max}$ (KBr) cm<sup>-1</sup>: 3369, 2923, 1659, 1609, 1498, 1449, 1357, 1304, 1277, 1212, 1176, and 1069. <sup>13</sup>C NMR spectral data (CD<sub>3</sub>OD, Table 1). <sup>1</sup>H NMR spectral data (CD<sub>3</sub>OD, Table 2). HR-SI-MS (negative) *m/z*: 771.2001 [M – H]<sup>-</sup> (calcd for C<sub>33</sub>H<sub>39</sub>O<sub>21</sub>, 771.1986).

Table 1.	<sup>13</sup> C NMR	spectral	data	for	compounds
1–3.					

$\delta_{\rm C}$	1	2	3
Aglycone			
2	159.3	159.2	159.1
3	134.8	134.9	135.1
4	179.6	179.6	179.8
5	163.1	163.1	163.1
6	100.1	100.1	100.1
7	166.1	166.2	166.1
8	95.0	95.0	94.9
9	158.7	158.6	158.5
10	105.8	105.7	105.7
1'	123.9	118.0	118.0
2'	132.4	146.0	145.9
3′	116.3	149.8	150.0
4′	161.5	116.3	116.3
5′	116.3	123.3	123.2
6′	132.4	123.3	123.0
Glc-1			
1″	101.1	100.1	101.0
2"	77.1	77.1	80.2
3″	82.3	82.7	74.8
4″	71.5	71.4	70.2
5″	77.9	77.9	75.3
6″	62.8	62.5	62.4
Glc-2			
1‴	104.6	104.9	104.9
2‴	75.5	75.5	75.3
3‴	78.3	78.2	78.2
4‴	71.5	71.3	71.2
5‴	78.0	78.0	78.0
6'''	68.3	68.2	67.2
Rha			
1////	102.3	102.2	102.0
2''''	72.4	72.4	72.4
3''''	72.2	72.2	72.2
4''''	74.0	74.0	74.0
5''''	69.8	69.8	69.8
6''''	17.9	17.9	17.9

Recorded at 125 MHz in CD<sub>3</sub>OD. All assignments have been confirmed by 2D techniques (H-HCOSY, HSQC, or HMBC).

#### 3.3.3 Compound 3

Yellow powder;  $[\alpha]_D^{25} - 65 (c = 0.43, MeOH)$ . UV  $\lambda_{max}$  (MeOH) nm: 255 and 353. IR  $\nu_{max}$ (KBr) cm<sup>-1</sup>: 3379, 2917, 1658, 1607, 1495, 1445, 1358, 1275, 1206, 1178, 1076, and 1039. <sup>13</sup>C NMR spectral data (CD<sub>3</sub>OD, Table 1). <sup>1</sup>H NMR spectral data (CD<sub>3</sub>OD, Table 2). HR-SI-MS (negative) *m/z*: 771.2006 [M – H]<sup>-</sup> (calcd for C<sub>33</sub>H<sub>39</sub>O<sub>21</sub>, 771.1986).

Table 2. <sup>1</sup>H NMR spectral data for compounds 1-3 ( $\delta$  in ppm and J in Hz).

$\delta_{ m H}$	1	2	3
A-ring			
6	6.19 d (2.0)	6.19 d (2.0)	6.18 d (2.0)
8	6.38 d (2.0)	6.37 d (2.0)	6.37 d (2.0)
B-ring			
2'	8.02 d (9.0)	7.65 d (2.0)	7.74 d (2.0)
3'	6.89 d (9.0)		
5'	6.89 d (9.0)	6.88 d (8.5)	6.88 d (8.5)
6′	8.02 d (9.0)	7.54 dd (2.0, 8.5)	7.55 dd (2.0, 8.5)
Glc-1			
1″	5.35 d (7.5)	5.26 d (8.0)	5.18 d (8.0)
2"	3.30 m, o	3.30 m, o	4.06, t (8.0)
3″	3.78 brd (2.0)	3.77 brd (2.4)	3.72 m
4″	3.31 m, o	3.32 m, o	3.80 brs
5″	3.58 m, o	3.58 m, o	3.61, t (6.0)
6″	3.80 m, o, 3.70 m, o	3.80 m, o, 3.72 m, o	3.80 m, o, 3.70 m, o
Glc-2			
1‴	4.75 d (7.0)	4.75 d (7.0)	4.75 d (5.5)
2′′′	3.37 m, o	3.39 m, o	3.38 m, o
3‴	3.30 m, o	3.32 m, o	3.35 m, o
4‴	3.39 m, o	3.40 m, o	3.40 m, o
5‴	3.41 m, o	3.43 m, o	3.41 m, o
6'''	3.78 brd (2.0), 3.31 m, o	3.78 m, 3.33 m, o	3.70 m, o, 3.38 m, o
Rha			
1////	4.47 d (1.0)	4.48 s	4.50 s
2''''	3.47 dd (9.5, 3.5)	3.48 dd (9.5, 3.5)	3.50 m, o
3''''	3.58 m, o	3.57 m, o	3.57 brs
4''''	3.23 m, o	3.24 t (9.0)	3.27 m, o
5''''	3.40 m, o	3.42 m, o	3.50 m, o
6''''	1.16 d (6.5)	1.18 d (6.5)	1.17 d (6.0)

Recorded at 500 MHz in CD<sub>3</sub>OD. All assignments have been confirmed by 2D techniques (H-HCOSY, HSQC, or HMBC). Coupling constants J in parentheses. o, overlapping with other signals.

#### 3.4 Acid hydrolysis and sugar analysis

A sample (1 mg) of 1-3 was dissolved in 1 ml of MeOH and loaded on a TLC (silica gel) plate. The plate was immersed with a solution of 8 ml of 10 N HCl at a temperature of 60°C for 20 min. The dried plate was loaded with standard sugars and chromatographed using *n*-BuOH-HOAc-H<sub>2</sub>O (4:1:2) system, then visualized with phenylamine-*ortho*-benzene-dicarboxylic acid reagent and [Glc ( $R_{\rm f} = 0.25$ ), Rha ( $R_{\rm f} = 0.45$ )].

## Acknowledgements

We are grateful to the financial support by a grant (2007AA10Z441) from the Chinese National Programs for High Technology Research and

Development, grants (07JC14025 and 07dz12055) from Science and Technology Commission of Shanghai Municipality, a grant (06AZ053) from Education Commission of Shanghai, and a grant (2005CCA04800) from the National Key Program for Base Research (973 program). We are also grateful to the Center of Analysis and Measurement, Institute of Chemistry, Chinese Academy of Sciences, for the HR-SI-MS; Laboratory of NMR Analysis and Measurement, South China Sea Institute of Oceanology, for the NMR spectra.

### References

- [1] N.A.G. Ramachandran and T.R. Seletharaman, *J. Nat. Prod.* **49**, 710 (1986).
- [2] C. Mahidol, H. Prawat, S. Ruchirawat, K. Lihkitwitwitayawuid, L.Z. Lin, and G.A. Cordell, *Phytochemistry* 45, 825 (1997).

- L.-R. Xu et al.
- [3] C. Mahidol, H. Prawat, W. Kaweetripob, and S. Ruchirawat, *Heterocyles* 57, 1287 (2002).
- [4] R.M. Narayana, G.L.D. Krupadanam, and G. Srimannarayana, *Phytochemistry* 37, 267 (1994).
- [5] L.R. Xu, J. Wu, and S. Zhang, J. Asian Nat. Prod. Res. 8, 9 (2006).
- [6] T. Sekine, M. Inagaki, F. Ikegami, Y. Fujii, and N. Ruangrungsi, *Phytochemistry* 52, 87 (1999).
- [7] M. Chuankamnerdkarn, S. Sutthivaiyakit, N. Thasana, and S. Pisutjaroenpong, *Hetero*cycles 57, 1901 (2002).
- [8] V. Rukachaisirikul, Y.S. Sukapondma, C. Jansakul, and W.C. Taylor, *Phytochemistry* 60, 827 (2002).
- [9] J. Takashima, N. Chiba, K. Yoneda, and A. Ohsaki, J. Nat. Prod. 65, 611 (2002).

- [10] M.C.D. Nascimento and W.B. Mors, *Phyto-chemistry* **11**, 3023 (1972).
- [11] M.C.D. Nascimento, R.L.D. Vasconcellos Dias, and W.B. Mors, *Phytochemistry* 15, 1553 (1976).
- [12] S. Sang, X. Cheng, N. Zhu, R.E. Stark, V. Badmaev, G. Ghai, R.T. Rosen, and C. Ho, *J. Agric. Food Chem.* **49**, 4478 (2001).
- [13] Y.P. Tang, Y.F. Li, J. Hu, and F.C. Lou, J. Asian Nat. Prod. Res. 4, 123 (2002).
- [14] G. Fico, A. Braca, A.R. Bilia, F. Tome, and I. Morelli, J. Nat. Prod. 63, 1563 (2000).
- [15] M.S. Kamel, K.M. Mohamed, H.A. Hassanean, K. Ohtani, R. Kasai, and K. Yamasaki, *Phytochemistry* 57, 1259 (2001).
- [16] J.H. Wang, F.C. Lou, Y.L. Wang, and Y.P. Tang, *Phytochemistry* 63, 463 (2003).